The Raman Fingerprint of Graphene

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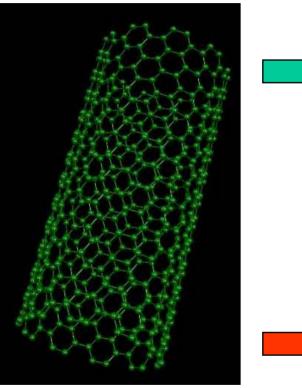
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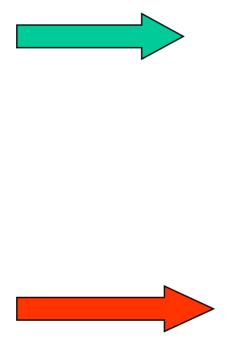


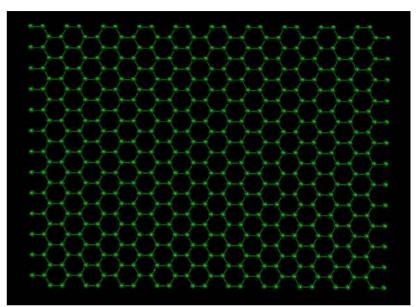


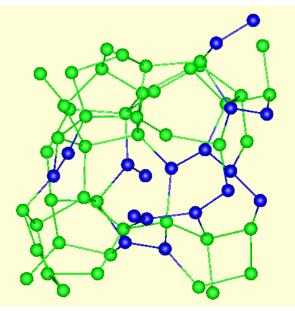


NanoTube-Evolution-Nagano 06









NanoTube-Evolution-Nagano 06

"Cut... Get graphene" Eklund-Sensei NT06

"Press (50GPa)...Get Diamond-like Carbon!"

S. Saito-Sensei CCNT06

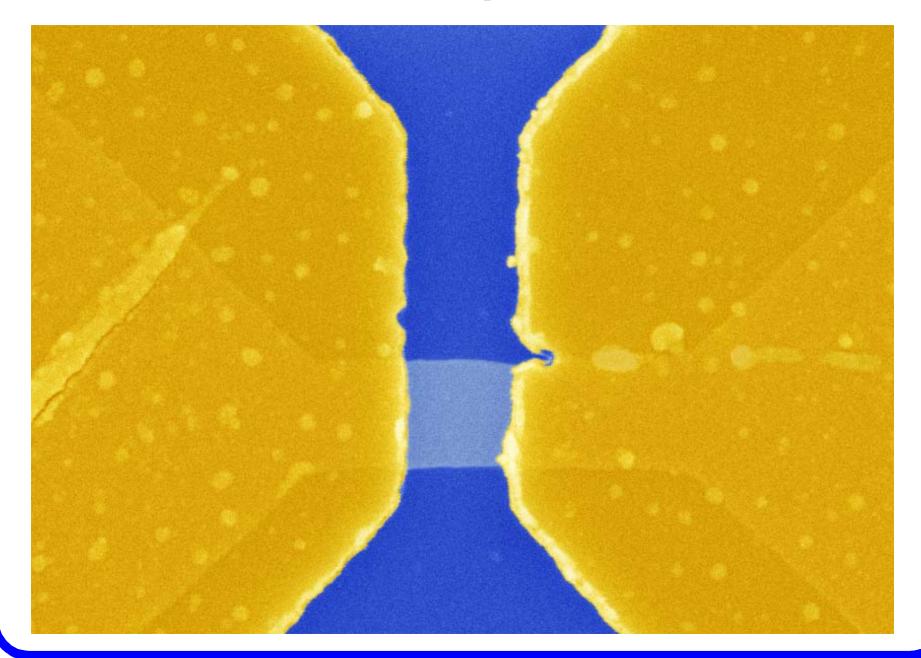
Graphenebig hype recently...

•Electron transport described by the (relativistic-like) Dirac equation Access to the rich and subtle physics of quantum electrodynamics in a relatively simple condensed matter experiment

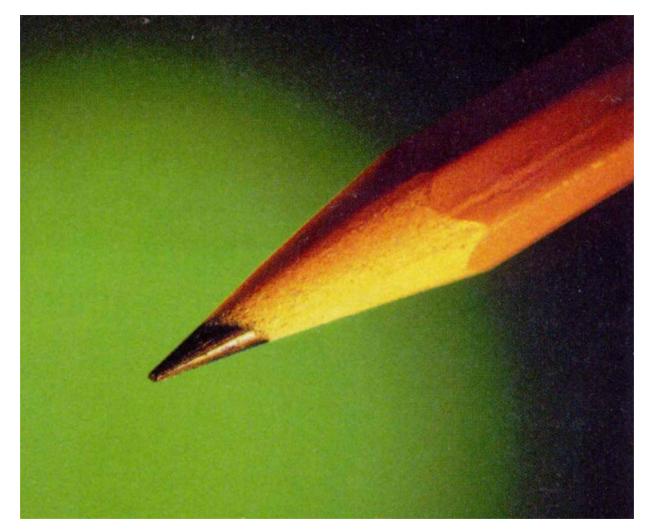
•Scalability of graphene devices to true nanometre dimensions makes it a promising candidate for future electronic applications, because of its ballistic transport at room temperature combined with chemical and mechanical stability.

•Graphene is the two-dimensional (2d) building block for carbon allotropes of every other dimensionality

Transistor: Graphene Ribbon

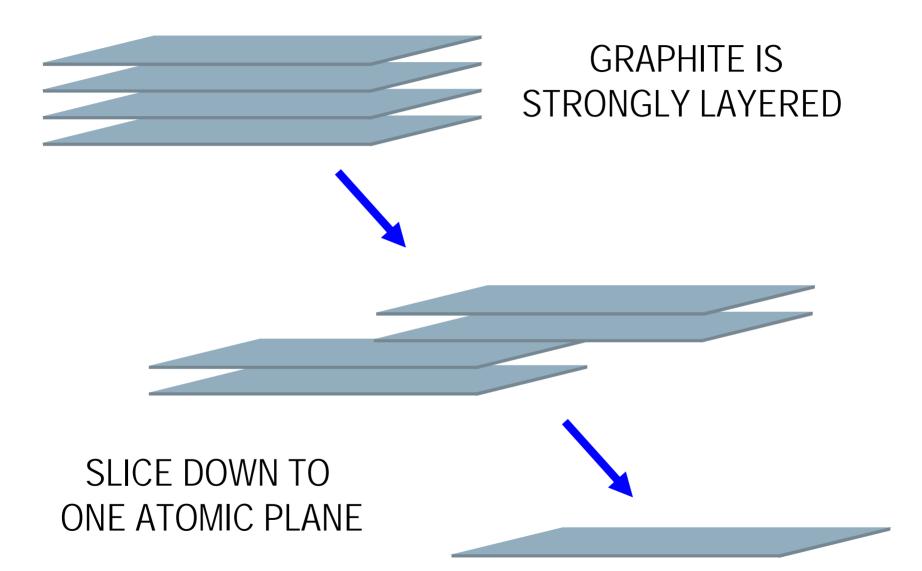


How to Make Graphene?



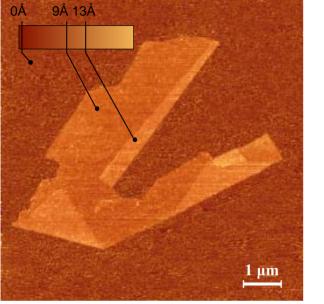
Drawing: (micro) mechanical cleavage of graphite

How to Make Graphene?



individual atomic sheets: do they exist?

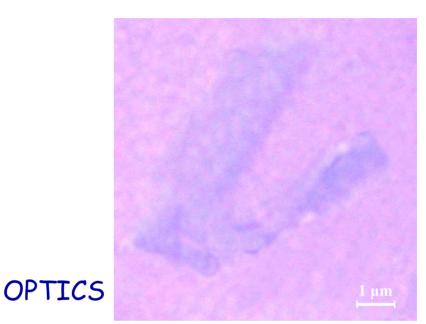
Free-Standing Graphene



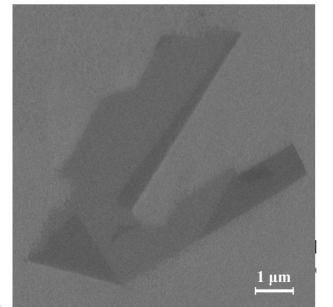
AFM

SEM

Key: Visual Identification

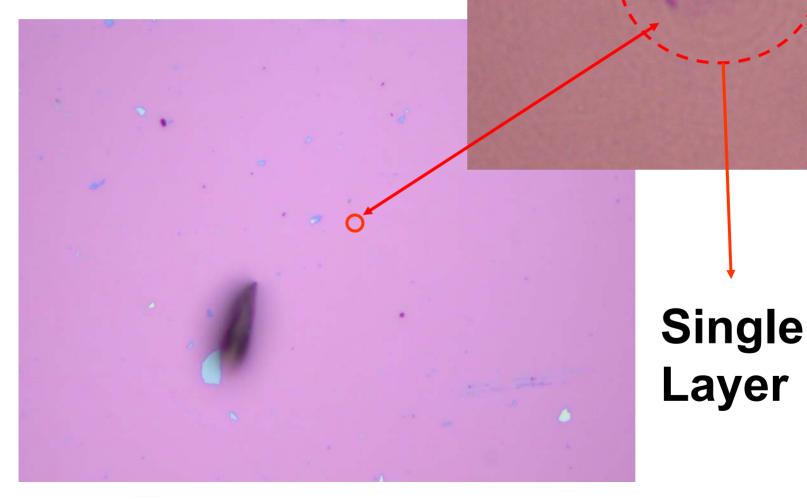


single layer of atoms visible by "naked" eye only on 300 nm SiO₂



However

Need Extremely Good Eye To Spot!!!





Two Layers



Mechanical cleavage is nice and simple However... Low yield, messy, not scalable

Better to grow graphene directly on substrate

This can be done...

But...not the subject of this talk



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Another Comment

AFM thickness of single layer is 0.5-1.5 nm! Due to chemical contrast

We want to be 150% sure

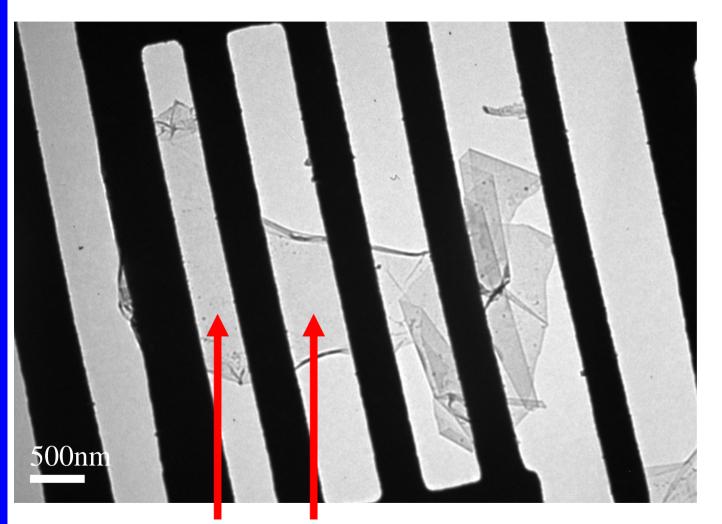
TEM



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Free-Hanging graphene sheets



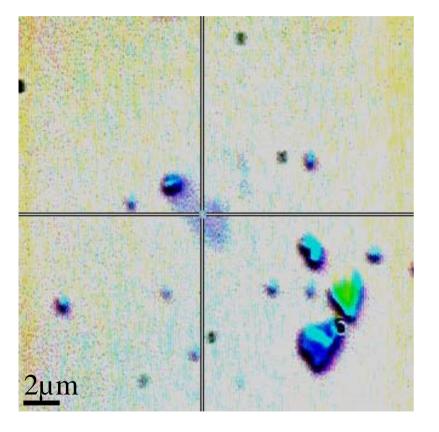
1 layer of graphene !

J. C. Meyer

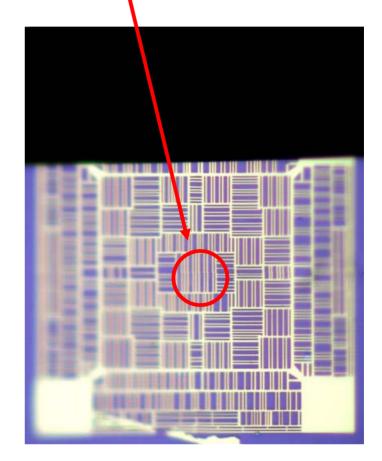


Preparation

1. Graphene sheet on substrate

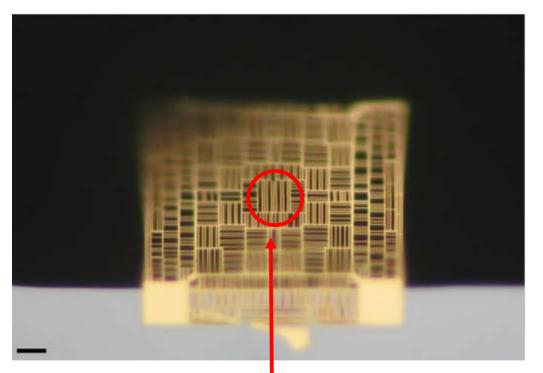


2. Metal grid patterned onto the flake



Preparation

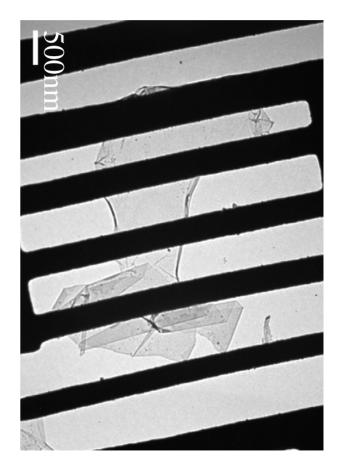
3. Etching of substrate



10µm

Flake remains in metal grid

4. TEM and electron diffraction analysis



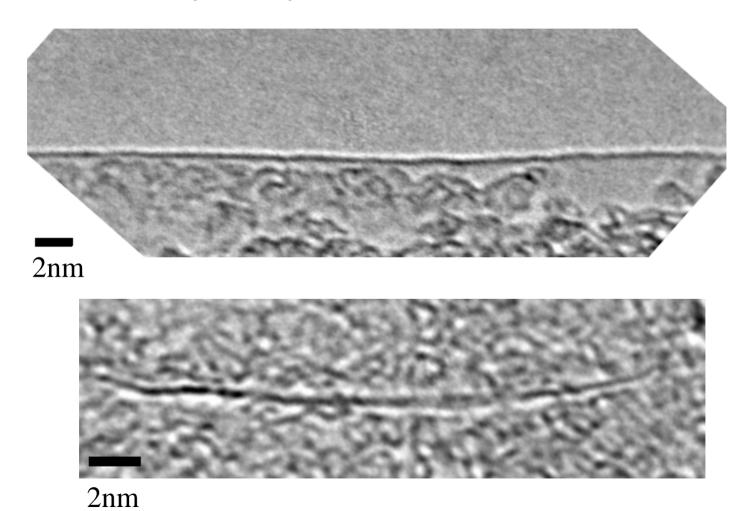
Electron diffraction:

Highly crystalline samples

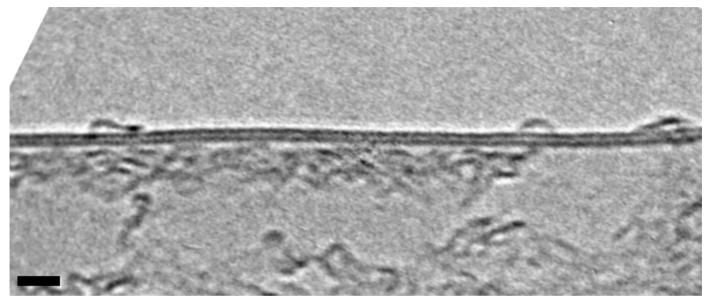
Number of layers? Diffraction tilt series!

One-layer graphene

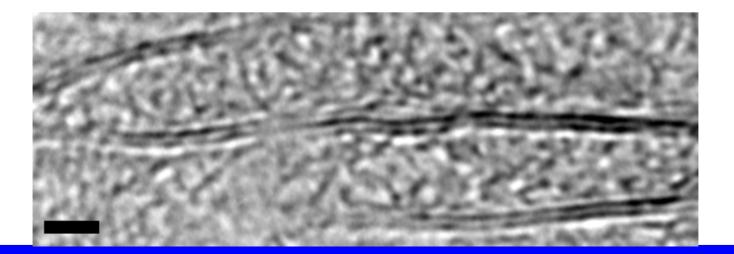
Sheets fold back at the edges, and sometimes show a wrinkle within the sheet. HRTEM analysis of the folding allows to verify the layer count.



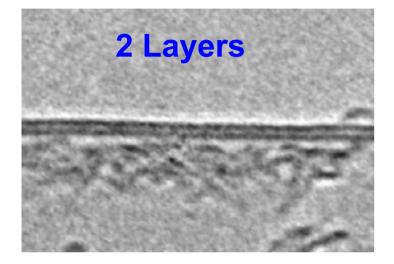
Two-layer graphene



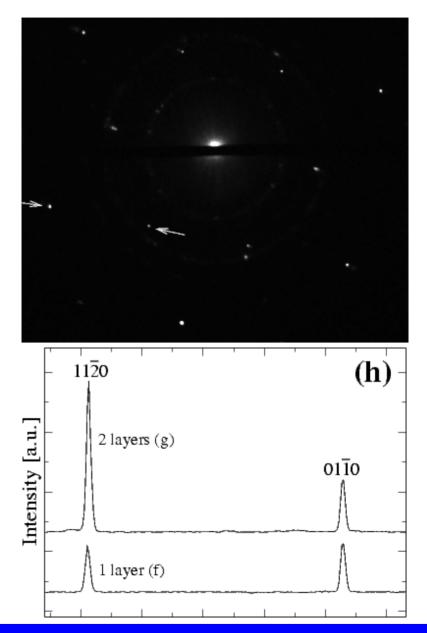
2nm



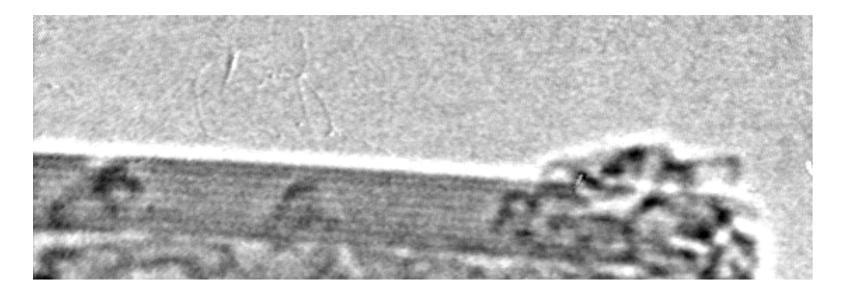
Two-Layer Graphene



Stacking A/B







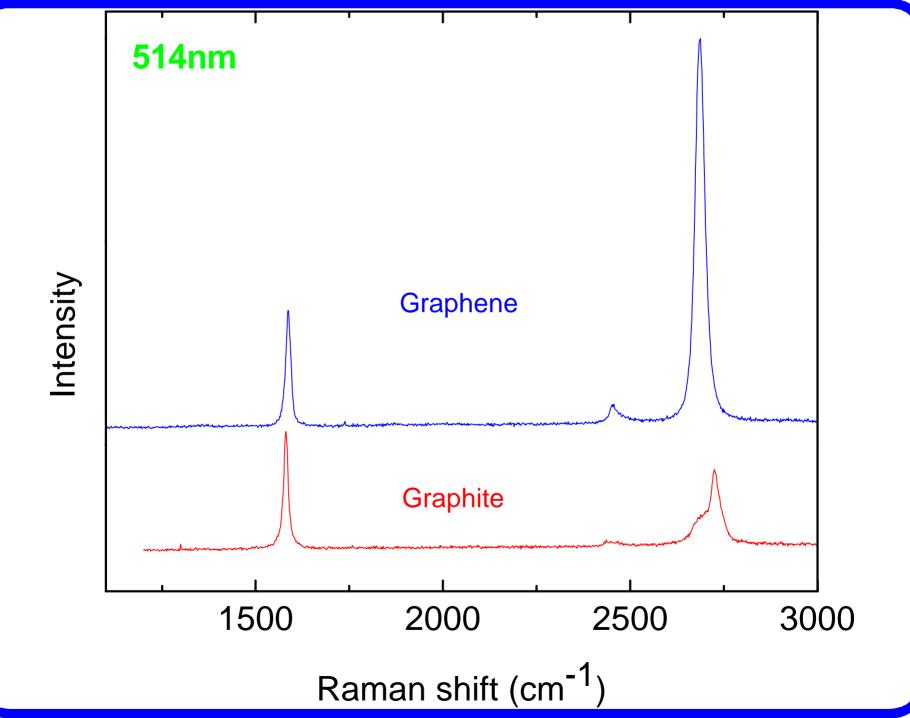


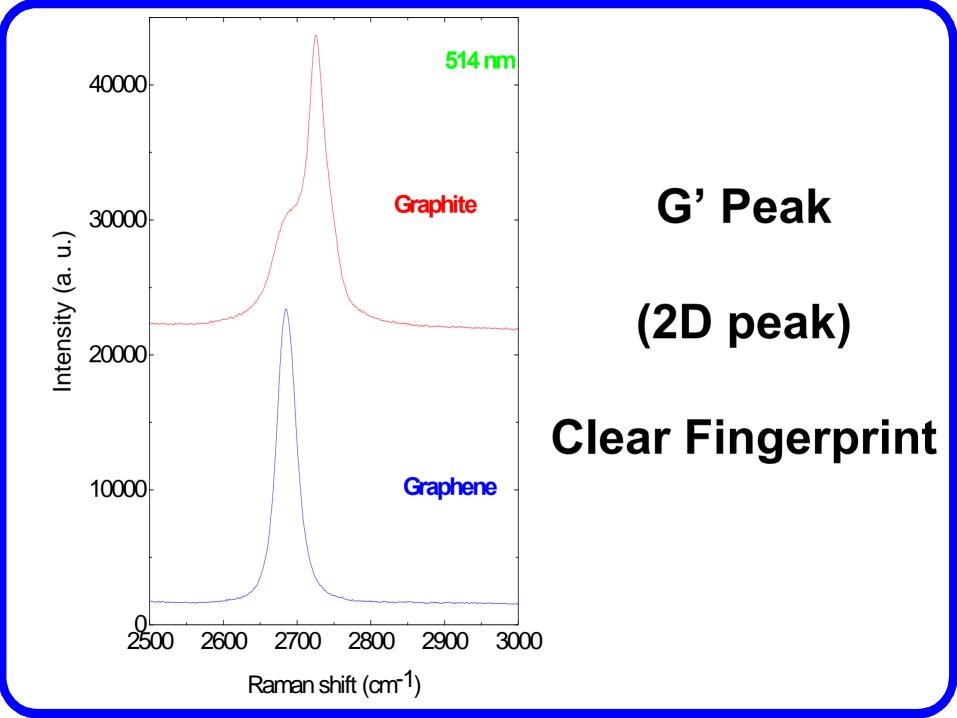
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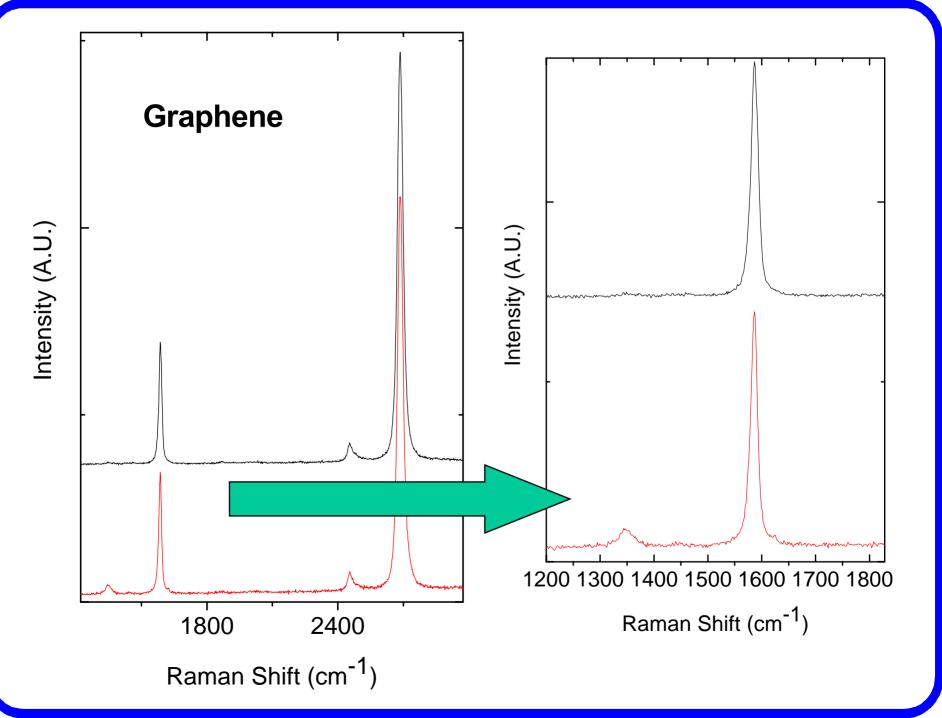


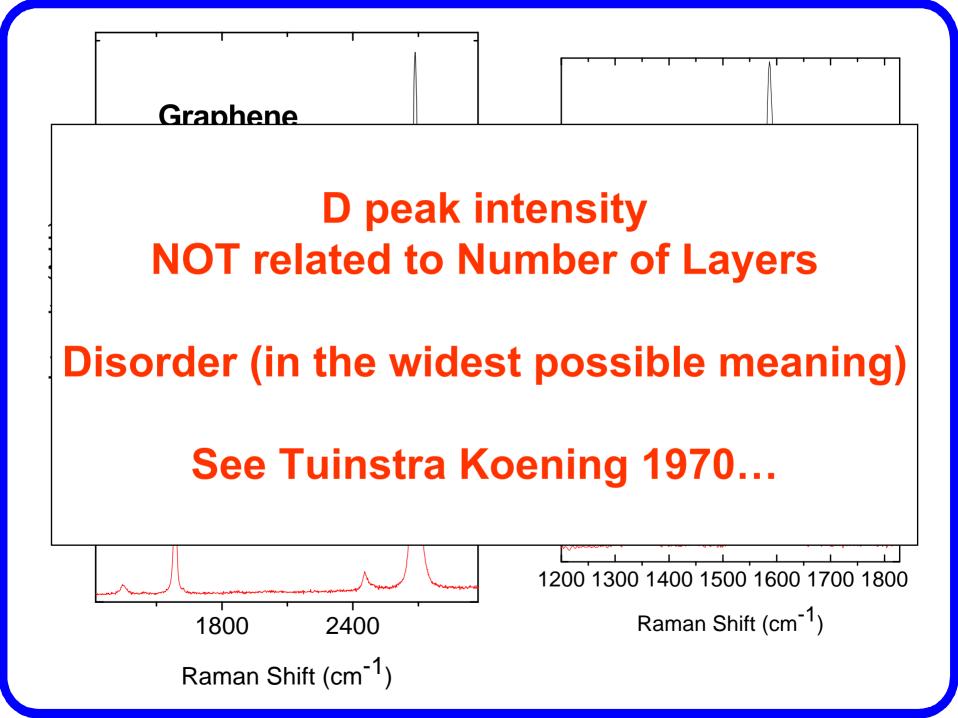
We Need High Throughput Non Destructive Quick Substrate Independent Identification Technique

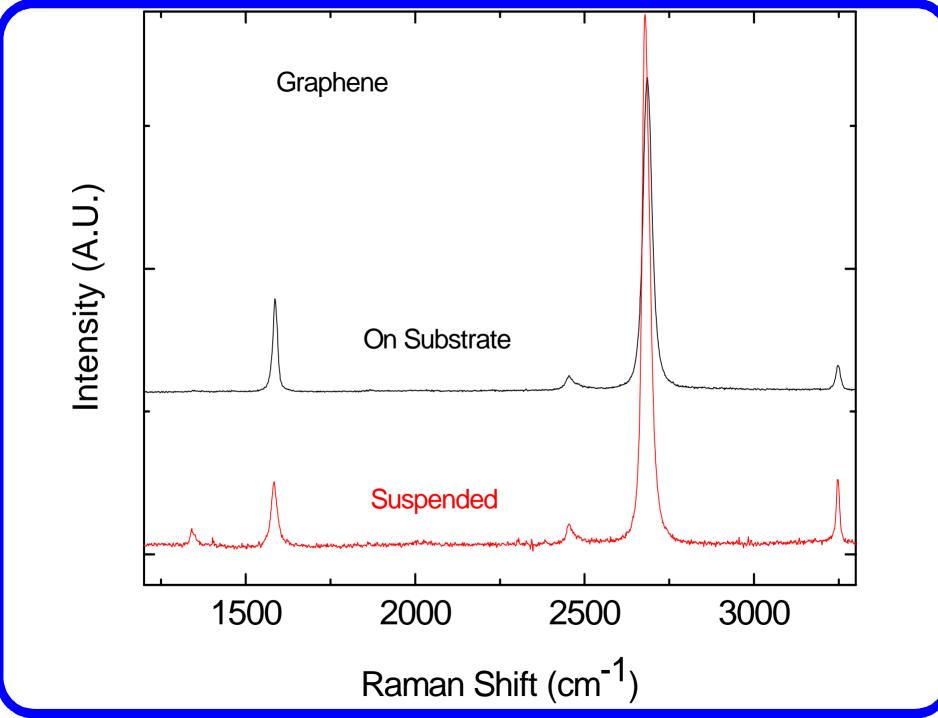
Raman Spectroscopy

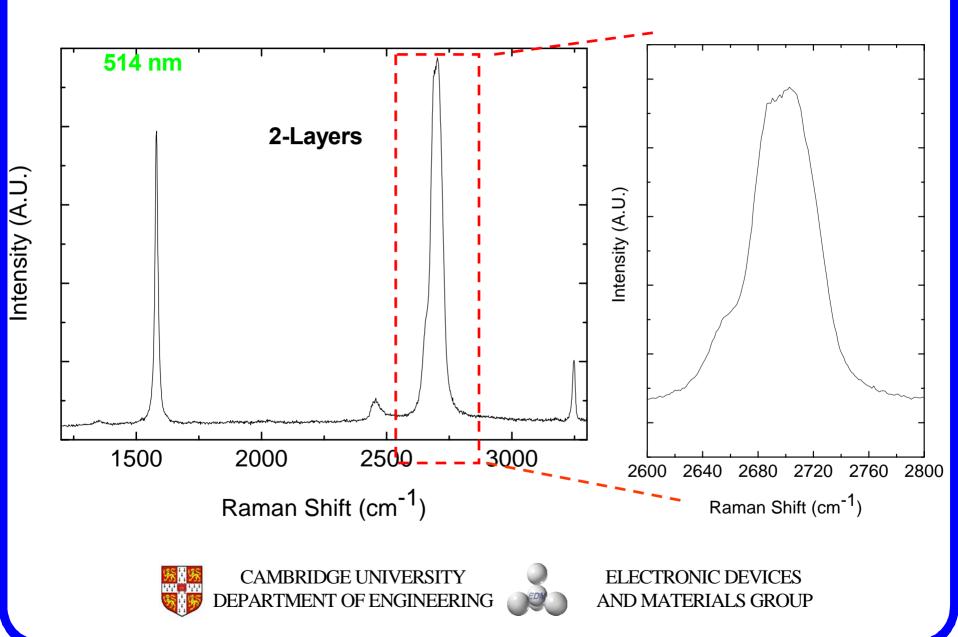




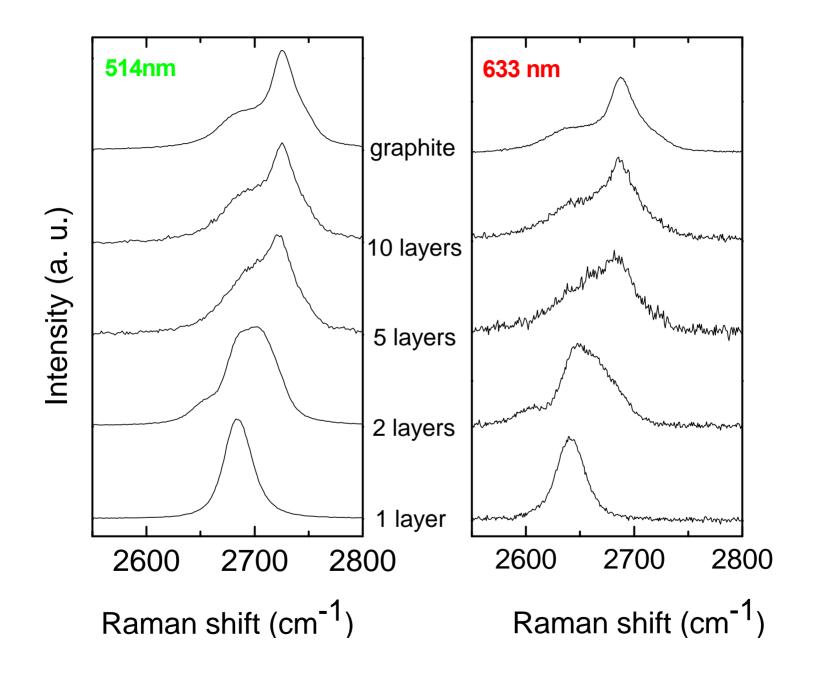


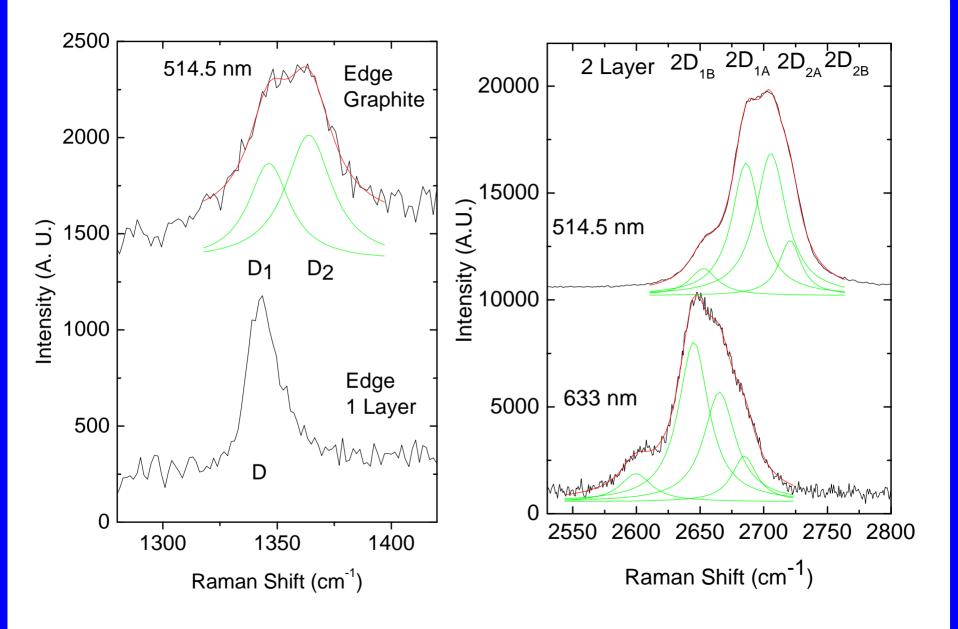






633 nm 514nm graphite Intensity (a. u.) 10 layers 5 layers 2 layers 1 layer WN. 1500 1550 1600 1650 1500 1550 1600 1650 Raman shift (cm⁻¹) Raman shift (cm⁻¹) Slight Upshift ~ 5cm⁻¹





The Origin of G' (2D) Peak

Named G' since is one of the 2 biggest peaks in graphite

BUT it is the second order of D peak

Nothing to do with G peak

D forbidden in perfect crystal By Raman Fundamental Selection Rule **q~0**

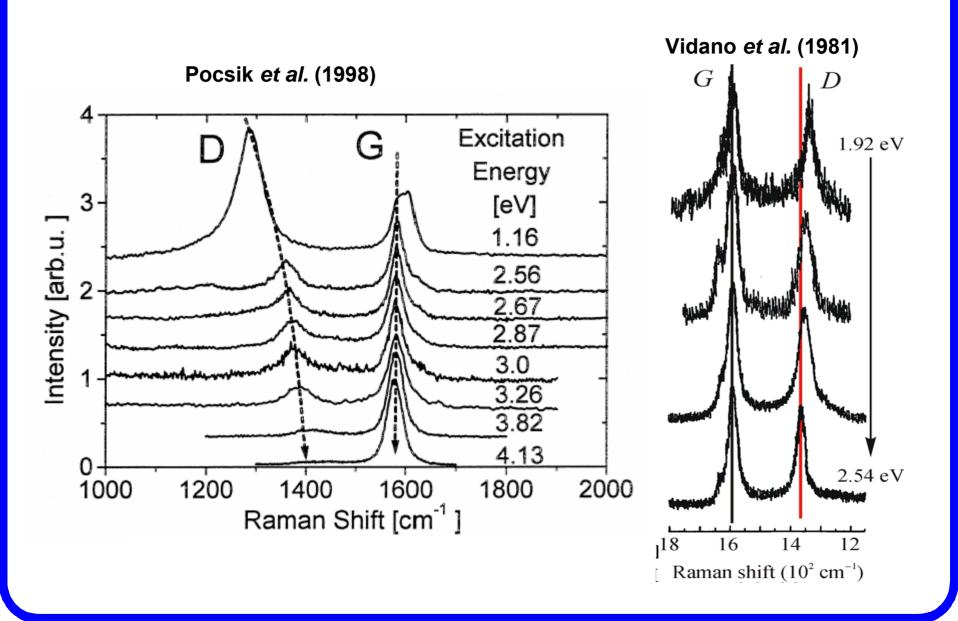
However 2nd order always allowed: q+(-q)=0



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D Peak Dispersion



D peak comes from LO phonons (Ferrari Robertson 2000)

Active by double resonance (Baranov 1988, Thomsen-Reich 2000)

Strongly dispersive due to Kohn Anomaly at K (Piscanec et al. 2004)



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Second order no defect scattering necessary

In principle ALL phonons active BUT

Double resonant phonons enhanced due to resonance and strong electron-phonon coupling



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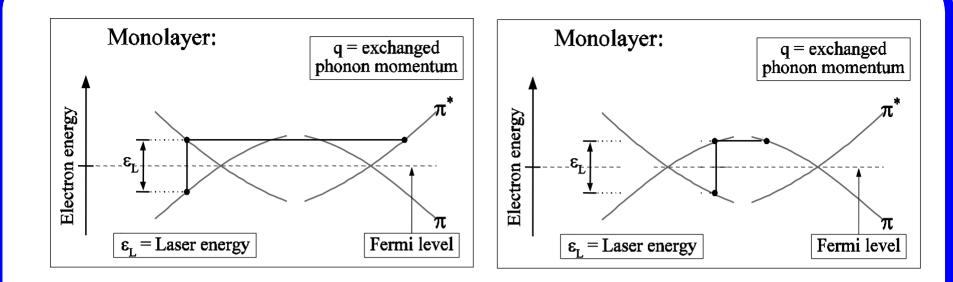


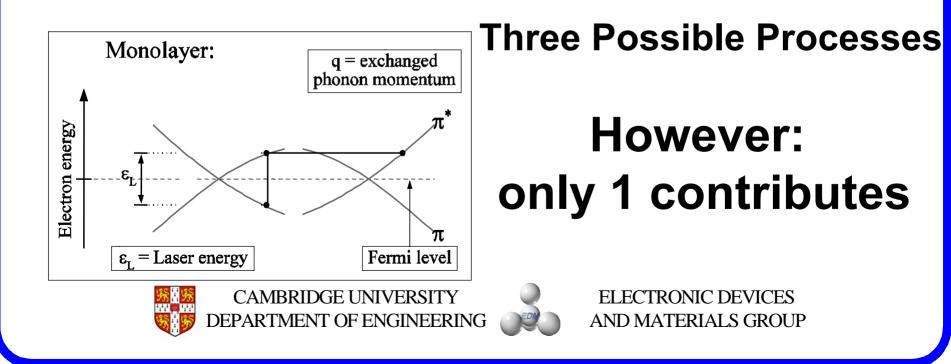
Previous double resonance models predict multiple D peaks for graphene in contrast with experiments

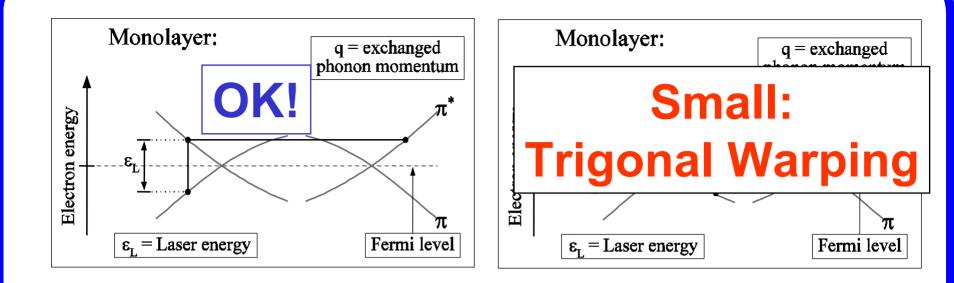
Double structure of 2D peak in graphite never explained

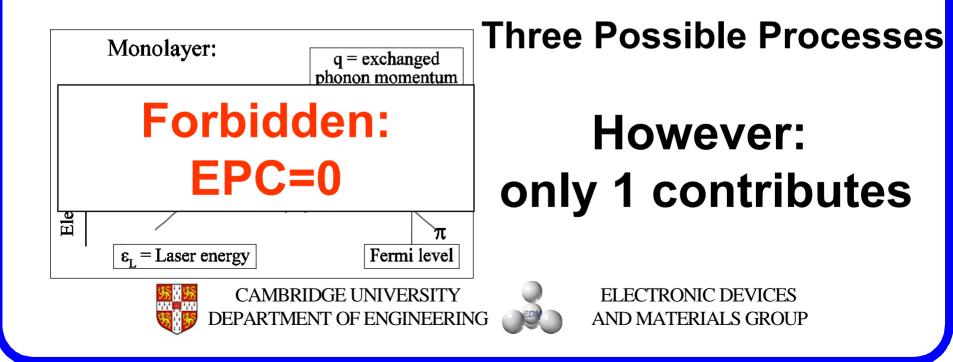
Traditional interpretation (1980) 2 Maxima in graphite Phonon Density of States at K and M WRONG since 2D disperses with excitation

KEY: Evolution of Electron Bands with number of layers

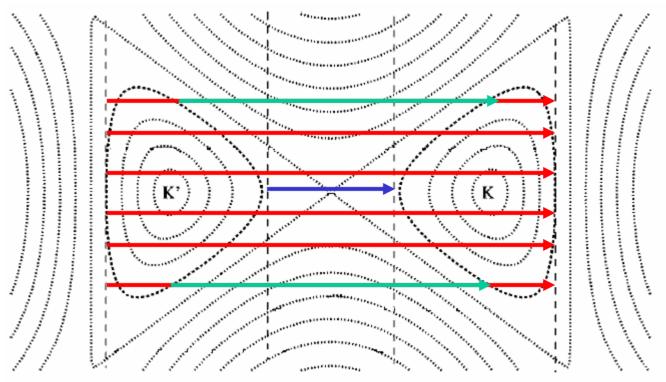








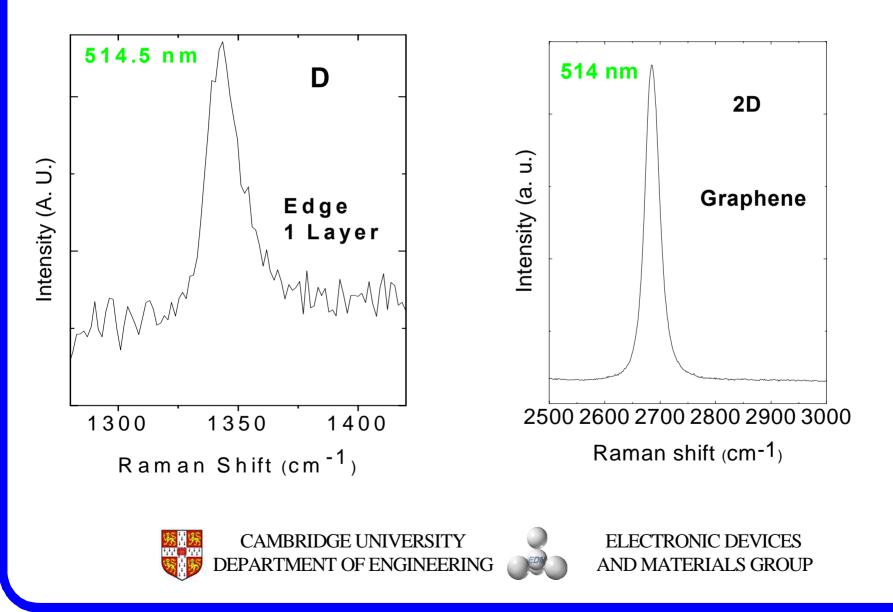
Trigonal Warping Effect



Adapted from: Kurti et al., Phys. Rev. B 65 165433 (2002)

q > K → Strong EPC and large portion of the phase-space
q < K → Strong EPC but small portion of the phase-space
q ~ K → EPC~0

1 Component D and 2D peaks



Two-layer Graphene

Two possibilities:

1) Phonon Splitting

Band splitting

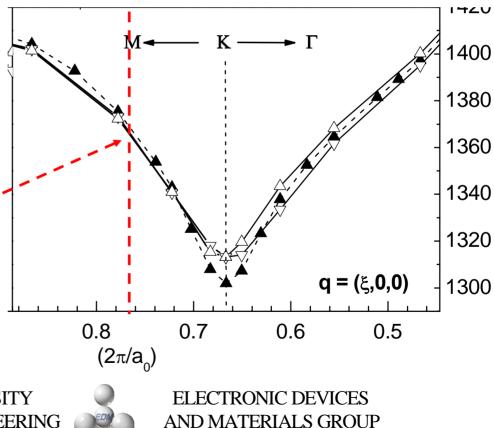
2)

Phonon Splitting K-M is Minor

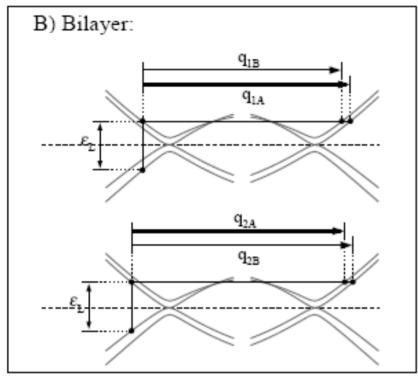
PRL 93, 185503 (2004)



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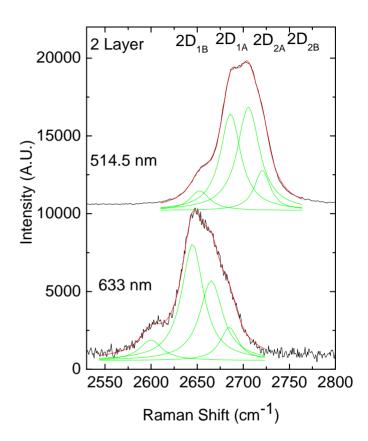


Band Splitting Main Effect





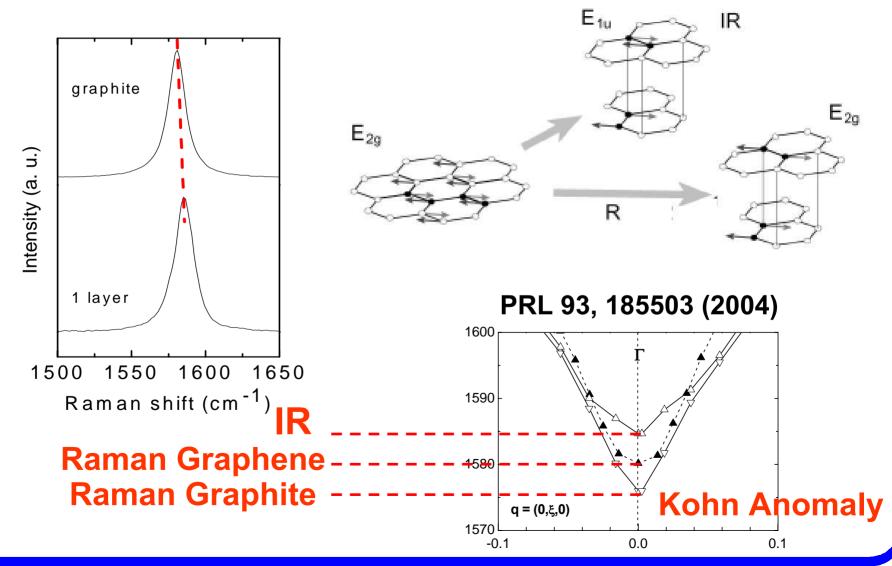
514.5	2 Layers				
Experimental	-44	-10	+10	+25	
Theory	-44	-11	+11	+41	
633					
Experimental	-55	-10	+10	+30	
Theory	-44	-9	+9	+41	



4 components 2 Most intense

Origin of Small Upshift of G peak

~5 cm⁻¹ Upshift



Phonon-Linewidths and EPC

In a perfect crystal, phonon linewidths determined by Interaction with other elementary excitations:

$$\gamma = \gamma^{an} + \gamma^{EPC}$$

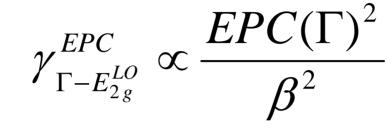
 γ^{an} : anharmonic contribution, due to interaction With other phonons. Determined by anharmonic terms in interatomic potential.

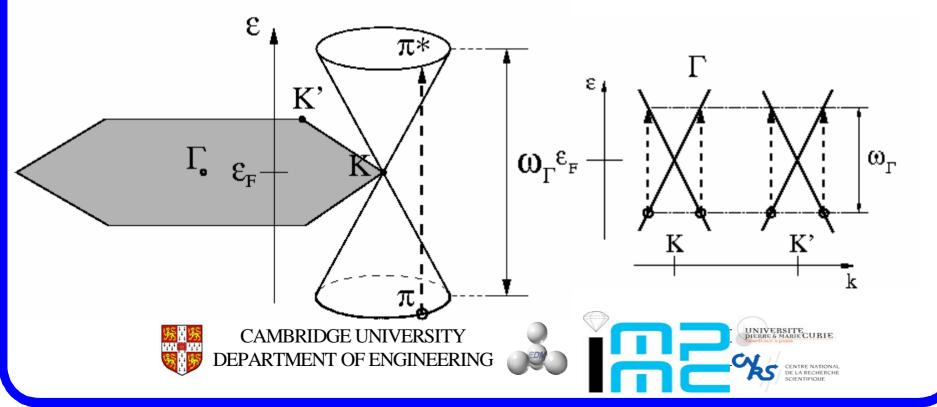
 γ^{EPC} : interaction with electron-hole pairs. Determined by EPC and present in systems with null electron gap

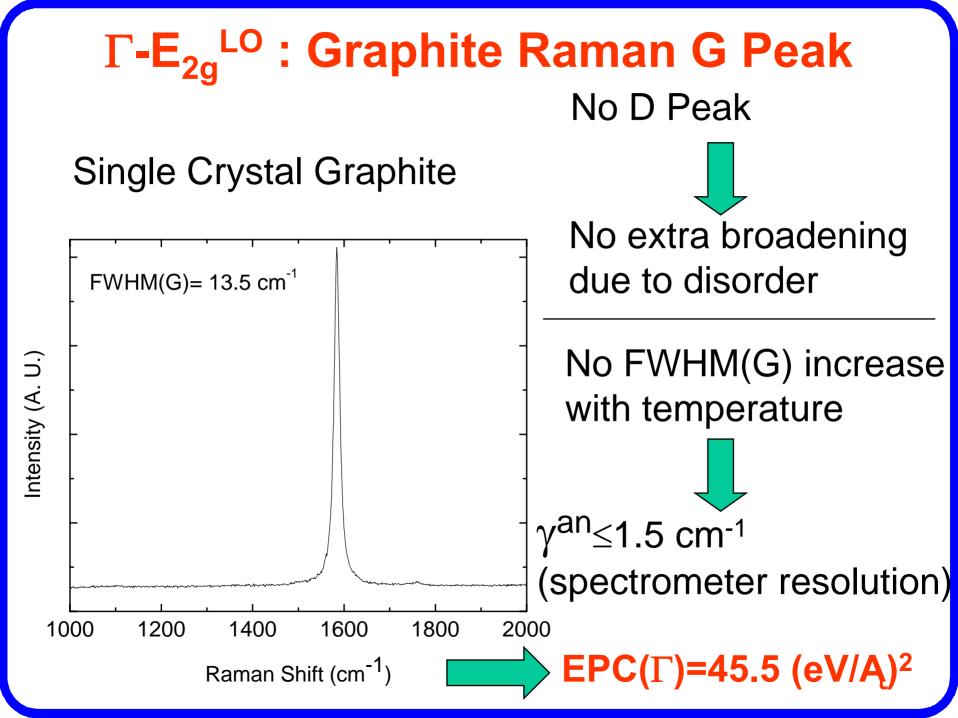




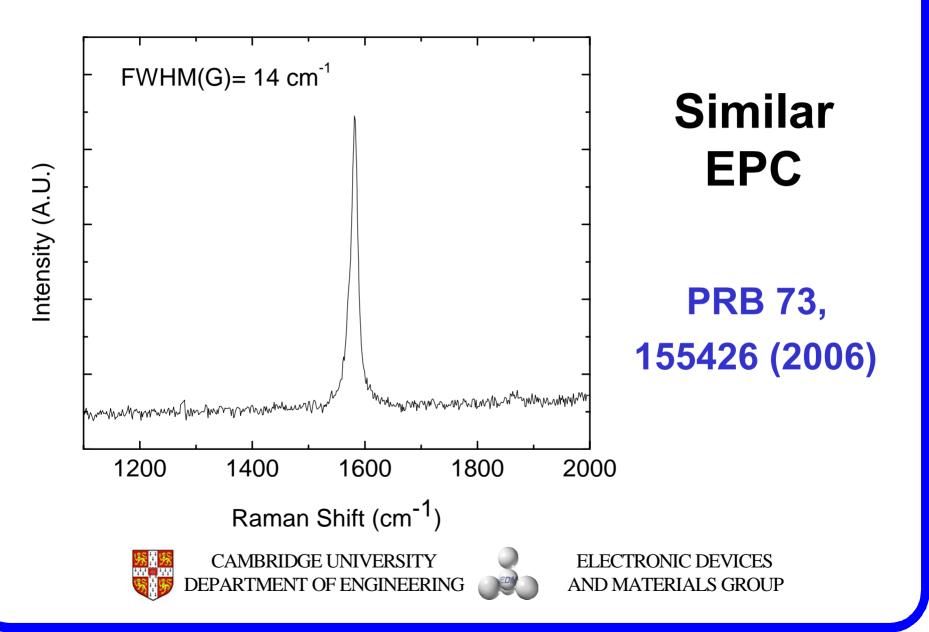
Γ-E_{2g}^{LO} Phonon Decay Processes From the Fermi Golden Rule:







And... Single layer graphene...



Implications for Nanotubes

Single 2D peak graphene \Rightarrow Single 2D peak in Single Wall CNT

Curvature and confinement give diameter dependence

2D(SWNT)~2D (graphene)- A/d

2D position in Graphite should not be used to scale

Distribution of SWNTs of different diameters, distribution of 2D peaks

What about Multi-Wall?

First approximation each wall gives a 2D peak

DWNT two 2D peaks (inner and outer wall)

HOWEVER, inter-wall interactions Can change simple picture Further splitting, Less peaks!

Details to follow...



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Conclusions

Identified unique features of Raman spectrum, which fingerprints graphene amongst all other carbon allotropes.

The Raman spectrum evolution with increasing number of layers reflects the evolution of the electronic structure and electron-phonon interactions

Raman spectroscopy is a quick, high-throughput, non-destructive technique for the unambiguous identification of graphene layers.

Raman+Graphene is Good Fun!



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The Leverhulme Trust









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CPU:

- EU FAMOUS & CANAPE project
- Marie Curie Fellowship IHP-HPMT-CT-2000-00209
- The Royal Society

Funding:

- EPSRC GR/S97620/01
- •The Leverhulme Trust



- HPCF, Cambridge UK
- IDRIS, Orsay France



Reference

arXiv:cond-mat/0606284

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 ¹Cambridge University, Engineering Department, Trumpington Street, Cambridge CB3 0FA, UK
 ² Max Planck Institute for Solid State Research, Stuttgart 70569, Germany
 ³Institut de Mineralogie et de Physique des Milieux Condenses, Paris cedex 05, France
 ⁴Department of Physics and Astronomy, University of Manchester, Manchester, M13 9PL, UK

Graphene is the two-dimensional (2d) building block for carbon allotropes of every other dimensionality. It can be stacked into 3d graphite, rolled into 1d nanotubes, or wrapped into 0d fullerenes. Its recent discovery in free state has finally provided the possibility to study experimentally its electronic and phonon properties. Here we show that graphene's electronic structure is uniquely captured in its Raman spectrum that clearly evolves with increasing number of layers. Raman fingerprints for single-, bi- and few-layer graphene reflect changes in the electronic structure and electron-phonon interactions and allow unambiguous, high-throughput, non-destructive identification of graphene layers, which is critically lacking in this emerging research area.



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